

PII: S0040-4020(96)00689-8

Reactivity of Condensed Thiophenes in the Diels-Alder Reaction: The Reactivity of 3-Aminothieno [3,4:3',4'] Benzo [b] Pyranone; 3-Aminothieno [3,4-c] Quinoline and of 5-Amino-7-Substituted Thieno [3,4-d] Pyridazinone Toward Electron-Poor Olefins and Acetylenes.

Fatima Al-Omran, Mervat Mohammed Abdel Khalik, Hanan Al-Awadhi and Mohammed Hilmy Elnagdi *

Department of Chemistry, Faculty of Science, University of Kuwait, P.O. Box 5969 Safat; 13060 Kuwait.

Abstract. The thieno [3,4:3',4'] benzo [b] pyranone (3b) and the thieno [3,4-c] quinoline (3c) are prepared via reacting 4-methylcoumarin-3-carbonitrile (9a) and 4-methyl-2-oxo-1,2-dihydroquinolin-3-carbonitrile (9b) with elemental sulphur. Similarly the thieno [3,4:3',4'] naphtho [1,2-b] pyran (11) is prepared from reaction of (10) and elemental sulphur. The condensed thiophenes (3b,c) and (11) react with a variety of electron-poor olefins to yield products of addition followed by hydrogen sulphide elimination. The reaction of (3b,c) and (11) with ethyl propiolate in refluxing dioxane/actic acid mixture affords the condensed thiepins (24a,c). The nature of product of reaction of (3b,c) and (11) with dimethyl acetylenedicarboxylate is dependent on applied reaction conditions. Thus the reaction of (3b,c) and (11) with dimethyl acetylenedicarboxylate in refluxing dioxane afforded condensed thiepins while at 250°C products of addition and desulfurization are obtained. Whereas thienopyridazinones (1a,b) react with ethyl acrylate and diethyl fumarate to yield thiadiazaacenaphthenones (31a-c) respectively, compounds (1a,b) react with N-phenylmaleimide to yield phthalazine (32). The thienopyridazinone (1c) failed to react with the same reagents under similar conditions. Copyright © 1996 Elsevier Science Ltd

Thiophenes are known to be highly inactive as 4π electron components in the Diels-Alder type [4+2] cycloaddition¹⁻³. However, Elnagdi et al⁴⁻⁸ have shown that 5-aminothieno[3,4-d]pyridazinones (1a,b) and some thieno [3,4-d] pyridines (2) react readily with electron-poor olefins to yield [4+2] cycloadducts that decompose under reaction conditions via loss of hydrogen sulfide or elimination of hydrogen molecule to yield benzoazines⁴⁻⁸. Also Gronowitz and Dahlgren⁹ have shown that benzo[c]thiophenes undergo ready 4+2 addition of dimethyl acetylenedicarboxylate to yield naphthalene derivatives. Recently, Döpp et al¹⁰ have reported that thieno [3,4:3',4'] benzo[b]pyranimine (3a) also adds diethyl fumarate and dimethyl acetylenedicarboxylate in Diels-Alder type reaction. While cycloadduct produced from reaction of (3a) with diethyl fumarate decomposed in a manner similar to that observed earlier by Elnagdi et al⁸ for (1a,b) and (2) producing (4), product of addition of (3a) to dimethyl acetylenedicarboxylate has rearranged into the thiepinobenzopyranone (5). We became thus interested to explore the behaviour of condensed thiophenes and to see whether a rearrangement into thiepins is a general reaction that can take place with other systems or with other reagents. Since participation of the imino function in (4) in addition reaction to the acetylene esters has resulted in the formation of mixtures of isomeric (5) that gave complex spectra which can not be used to exclude completely other possible structures for the formed thiepins, we have decided to work with

2-oxobenzopyranone and not with the imines. In the present paper we report synthetic approaches to several new condensed aminothiophenes as well as on their behaviour towards electron-poor olefins and acetylenes. Thus (1a,b) was prepared utilizing our newly reported synthetic approach. Compound (1c) was prepared via coupling in situ generated 4-phenyl-3-oxobutanitrile (6) with benzenediazonium chloride and subsequent condensation of the produced phenylhydrazone (7) with ethyl cyanoacetate. The formed pyridazinone (8) then reacted readily with sulphur to yield (1c).

Treatment of 2-hydroxyacetophenone with ethyl cyanoacetate produced (9a). The latter compound reacted with sulphur in refluxing DMF/piperidine, affording thieno[3,4:3',4'] benzo[b]pyran (3b) in good yield. Similarly condensing 2-aminoacetophenone with ethyl cyanoacetate produced (9b) which reacted with sulphur

to yield (3c). In a similar way condensing 2-acetyl-1-naphthol with ethyl cyanoacetate gave (10) which reacted with elemental sulphur to yield (11). The formation of (3b,c) and (11) from (9a,b) and (10) is further extension of our condensed thiophenes synthesis from alkylhetero-aromatic carbonitriles and sulphur.

Compounds (3b) and (11) reacted with diethyl fumarate at 250°C to yield products of addition and hydrogen sulphide elimination which are formulated as (12a) and (13a) respectively.

$$\begin{array}{c} CO_2R \\ CO_2R \\ NH_2 \\ \end{array}$$

$$\begin{array}{c} 12 \text{ a,} X=O; R=Et \\ \text{b,} X=O; R=Me \\ \text{c,} X=NH; R=Me \\ \end{array}$$

$$\begin{array}{c} 13 \text{ a,} R=Et \\ \text{b,} R=Me \\ \end{array}$$

It is assumed that initially formed cycloadducts (14) first rearrange into either (15) or (17) and then transfer a proton yielding (16) that aromatize into final isolable products. Although intermediacy of (15) seems most likely as it is more stable than (17), involvement of (17) cannot be ruled out, as adducts produced by Döpp et al⁹ can rearrange only via similar intermediate into the isolated thiepins.

The reaction of (3b,c) with dimethyl acetylenedicarboxylate and with ethyl propiolate in refluxing dioxane has resulted in formation of 1:1 adducts. These can thus be formulated as cycloadducts (18) or isomers (21-24). Thus ring opening of (18) would lead to (19) or (20). Species (19) can then isomerise either to (21) or (23) while (20) can also give (22) or (24).

$$\begin{array}{c|c}
CO_2Et \\
CO_2Et \\
NH_2
\end{array}$$

$$\begin{array}{c|c}
CO_2Et \\
NH_2
\end{array}$$

$$\begin{array}{c|c}
CO_2Et \\
NH_2
\end{array}$$

$$\begin{array}{c|c}
CO_2Et \\
-H_2S
\end{array}$$

$$\begin{array}{c|c}
17
\end{array}$$

Structures (18, 21 or 22) were readily ruled out based on ¹³C NMR of the products which revealed absence of signals for sp³ carbons other than the two methoxy carbons. Stucture (23) was also excluded based on NOE difference experiments which has revealed structure (24a). Thus, irradiation of the low field signal at δ 8.23 ppm in (24a) enhanced the doublet at δ 5.90 ppm as well as benzene ring protons at δ 7.93 ppm. When compounds (24b,d) were heated above their melting points in absence of a solvent for short period, compounds (12b,c) were obtained. The formation of (12b,c) from (24b,d) is assumed to occur via $\delta \pi$ -electrocyclization of (24b,d) to yield (22) which on desulphurization afforded (12b,c). Similar to the behaviour of (3b,c), compound (11) also reacted with dimethyl acetylenedicarboxylate in refluxing dioxane to yield the thiepin (25). Although thiepins are predicted to have anti-aromatic character¹¹ and are expected to undergo ready sulphur extrusion¹², reports on some annulated thiepins¹³⁻¹⁶ have ascribed their thermal stability to delocalisation of the thiepin π -electrons into the π -electron system of annulated rings giving rise to azulene-like charge separated structures which is possible in our systems.

When the reaction of (3b) and (11) was conducted at 250°C, the reaction produced (12b) and (13b) respectively via addition and desulfurization. Attempted preparation of (12c) similarly failed. We believe that thiepins (24 and 25) are intermediates in these reactions and that under reaction conditions they undergo electrocyclisations yielding (22) which are decomposed further into isolated products.

Similar to their behaviour toward diethyl fumarate compounds (3b,c) and (11) also added N-phenylmaleimide to yield (26a,b) and (27) respectively.

Recently we have reported that (1a) reacts with acrylonitrile to yield the thiadiazaacenaphthenes (31a) whose structure was confirmed by X-ray crystallography⁸. Now we report that also (1a) reacts with ethyl acrylate and (1b) with diethyl fumarate to yield the thiadiazaacenaphthenes (31b,c). These are formed via intermediacy of (28) that ring opens into (30) then lose a proton and cyclise into final isolable products via elimination of methoxide ion. It is thus confirmed that intermediates like (29) are not formed in reaction of (1a,b) with electron-poor olefins although the presence of methyl function renders such intermediate more stable than intermediate that led to products isolated by Döpp et al¹⁰ as well as those obtained here. Fusion of (1a) with N-phenylmaleimide at 140°C produced the phthalazine (32). It seems that under such condition, loss of hydrogen sulphide and not methanol elimination is the predominating reaction pathway. Compound (1c) failed to react with either dimethyl acetylenedicarboxylate or with electron-poor olefins under a variety of conditions. Although the introduction of a phenyl substituent decreases the HOMO-LUMO energy of the system and thus is expected to increase reaction rate, also the phenyl substituent sterically hinders the approaching reagent.

In conclusion, amino condensed thiophenes are active as 4π electron systems. With acetylenes the formed cycloadducts undergo facile opening by cleavage of the C-4, sulphur bond to yield products whose structure depends upon the reagent and reaction conditions. These, when heated (~250°C), undergo sulphur elimination via initial electrocyclisation into condensed thiranes giving dibenzopyrans, benzoquinolines and benzonaphthopyrans. With olefins, the cycloadducts undergo ring opening and then lose hydrogen sulphide affording (12-13).

Acknowledgment:

This work is financed by University of Kuwait research grant SC 071. We are grateful to the University of Kuwait general facility projects in the Chemistry Department for the analytical and spectral measurements.

Experimental

All melting points are uncorrected. IR spectra were recorded in KBr disks using a Shimadzu IR-740 spectrophotometer. ¹H and ¹³C NMR spectra were recorded on a Bruker AC-80 spectrometer with [D₆] DMSO as solvent and TMS as internal standards; chemical shifts are reported in δ units (ppm). Mass spectra were measured on Gs/MS INCOS XL Finnigan MAT. Microanalysis were performed on LECO CHNS-932. Compounds (1a,b) were prepared following our recently published procedure⁸.

3-Oxo-4-phenyl-2-phenylhydrazonobutanonitrile (7): A solution of acetonitrile (4.1g, 0.1mol) in toluene (100 mL) was treated with ethyl phenylacetate (16.4g, 0.1mol), then sodium hydride was added (4.0g, 60%). The reaction mixture was refluxed for 1.5h then evaporated in vacuo. The remaining solid product was triturated with water, neutralized with hydrochloric acid and extracted with chloroform. Evaporation of the chloroform layer afforded an oily syrup which was dissolved in ethanol (100 mL) and treated with sodium acetate (10g), then gradually treated under stirring with a solution of benzendiazonium chloride (prepared from 9.3g,0.1mol of aniline and appropriate quantities of both hydrochloric acid and sodium nitrite as has been described earlier⁸). The solid product, so formed, was collected by filtration and crystallized from ethanol. Compound 7 was obtained as yellow crystals from ethanol (70%); m.p. 137°C; IR: 3235 (NH), 2205 (CN) and 1681cm⁻¹ (CO). ¹H NMR: δ 4.20 (s, 2H, CH₂), 7.27-7.56 (m, 10H, 2Ar-H) and 12.30 ppm (br, 1H, NH); Anal. Calcd. for C₁₆H₁₃N₃O (263.29): C,72.98; H,4.98; N,15.96. Found: C,72.97;H,5.04; N,15.99.

4-Benzyl-1,6-dihydro-6-oxo-1-phenylpyridazine-3,5-dicarbonitrile (8): A mixture of 7 (2.63g, 0.01mol), ethyl cyanoacetate (1.13g, 0.01mol), ammonium acetate (3.0g) and acetic acid (0.6 mL) was heated with stirring at 200°C for half an hour, then left to cool and triturated with ethanol. The solid product, so formed, was collected by filtration and crystallized from ethanol. Compound 8 was obtained as yellow crystals (70%) m.p. 212°C, IR: 2225(CN) and 1673 cm⁻¹ (ring CO). ¹H NMR: δ 4.27 (s, 2H, CH₂), 7.39-7.57 ppm (m, 10H, 2Ar-H); ¹³C NMR: δ 151.78 (C-6), 139.69 (C-3), 134.22 (C-4), 133.26, 129.80, 129.04, 128.62, 127.80, 126.98, 125.51 and 121.64 (ring carbons), 115.07 and 113.08 (2CN) and 112.55 (C-5); MS: m/z (EI) 313 (M+1); Anal. Calcd. for $C_{19}H_{12}N_4O$ (312.32): $C_{19}C_{1$

5-Amino-3,4-dihydro-3,7-diphenyl-4-oxo-thieno[3,4-d]pyridazine-1-carbonitrile (1c): A suspension of compound 8 (3.12g, 0.01mol), in dioxane (10 mL) was treated with elemental sulphur (0.32g, 0.01mol) and piperidine (0.2 mL). The reaction mixture was refluxed for 4h, then poured onto water, the solid product, so formed, was collected by filtration and crystallized from the proper solvent. Compound 1c was obtained as red crystals from dioxane (82%) m.p. 220°C; IR: 3400 and 3295 (NH₂), 2220(CN) and 1656 cm⁻¹ (ring CO). ¹ H NMR: δ 7.40-7.50 (m, 10H, 2Ar-H) and 7.88 ppm (br, 2H, NH₂); ¹³C NMR: δ 162.85 (C-4), 157.85 (C-1), 140.24 (C-5), 130.59, 129.99, 128.68, 128.23, 127.98, 127.32, 125.51 and 121.29 (ring carbons),119.14 (C-

7), 118.28 (CN), 113.37 (C-4a) and 104.84 (C-7a); MS: m/z (EI) 345 (M+1); Anal. Calcd. for $C_{19}H_{12}N_4OS$ (344.32): $C_{19}H_{12}H_{12}H_{12}H_{12}H_{13}H$

General procedure for the preparation of (9a,b) and (10): A mixture of ethyl cyanoacetate (1.13g, 0.01mol) and ammonium acetate (5g) was treated with either of o-hydroxyacetophenone, o-aminoacetophenone or 2-acetyl-1-naphthol (0.01mol). The reaction mixture was heated at 220°C for 15min, left to cool and then triturated with ethanol. The solid product, so formed, was collected by filtration and crystallized from the proper solvent.

4-Methyl-2-oxo-benzo[b|pyran-3-carbonitrile (9a): Compound 9a was obtained as yellow crystals from dioxane (80%); m.p. 197°C; IR: 2220(CN) and 1714cm⁻¹ (ring CO). ¹H NMR: δ 2.73 (s, 3H, CH₃) and 7.41-8.05 ppm (m, 4H, Ar-H); ¹³C NMR: δ 164.14 (C-2), 161.15 (C-8a), 157.27 (C-4a), 153.99 (C-4), 135.82 (C-7), 127.61 (C-6), 126.03 (C-5), 119.15 (C-8), 117.71 (CN), 102.50 (C-3) and 18.28 (CH₃); MS: m/z (EI) 185 (M⁺); Anal. Calcd. for C₁₁H₇NO₂ (185.18):C,71.34; H,3.80; N,7.56. Found: C,71.36; H,3.83; N,7.55.

4-Methyl-2-oxo-1,2-dihydroquinoline-3-carbonitrile (9b): Compound 9b was obtained as white crystals from dimethylformamide (78%) m.p. 320°C; IR: , 3465 (NH), 2210(CN) and 1648 cm⁻¹ (ring CO). ¹H NMR: δ 2.71 (s, 3H, CH₃), 7.28-7.93 (m, 5H, Ar-H) and 12.20 ppm (br, 1H, NH); ¹³C NMR: δ 158.87 (C-2), 158.80 (C-8a), 139.43 (C-4), 133.96 (C-4a), 126.75 (C-7), 123.16 (C-5), 118.36 (C-6), 116.47 (C-8), 115.86 (CN), 106.35 (C-3) and 18.47 (CH₃); MS: m/z (EI) 184 (M⁺); Anal. Calcd. for $C_{11}H_8N_2O$ (184.19): C,71.73; H,4.37; N,15.20. Found: C,71.78; H,4.37; N,15.27.

4-Methyl-2-oxo-naphtho[1,2-b]pyran-3-carbonitrile (10): Compound 10 was obtained as light green crystals from dimethylformamide (70%) m.p. 280°C; IR: 2220 (CN) and 1724 cm⁻¹ (ring CO). ¹H NMR: δ 2.81 (s, 3H, CH₃) and 7.79-8.04 ppm (m, 6H, Ar-H); ¹³C NMR: δ 163.47 (C-2); 161.33 (C-10b); 136.47, 130.59, 128.42, 128.40, 128.27, 125.40, 122.77, 121.63, 121.36 (ring carbons), 119.92 (CN), 114.90 (C-4), 110.90 (C-3) and 18.75 (CH₃); MS: m/z (EI) 235 (M⁺); Anal. Calcd. for $C_{15}H_9NO_2$ (235.24): C,76.58; H,3.85; N,5.95.Found: C,76.96; H,3.95; N,5.73.

Reaction of (9a,b) and (10) with elemental sulphur: A solution of each of 9a,b and of 10 (0.01mol) in dimethyl formamide (10 mL) was treated with elemental sulphur (0.32g, 0.01mol) and piperidine (0.2 mL). The reaction mixture was refluxed for 4h, then poured into water, the solid product, so formed, was collected by filtration and crystallized from dioxane / ethanol mixture.

3-Aminothieno[3,4:3',4']benzo[b]pyran-4-one 3b: Compound 3b was obtained as brown crystals (86%) m.p. 250°C; IR: 3445 and 3335 (NH₂) and 1682 cm⁻¹ (ring CO). ¹H NMR: δ 6.89 (s, 1H, 1-H) and 7.10-7.94

ppm (m, 6H, Ar-H and NH₂); ¹³C NMR: δ 166.80 (C-4), 151.27 (C-5a), 131.18 (C-3), 129.38 (C-9a), 124.50 (C-7), 124.30 (C-9), 124.03 (C-1), 118.30 (C-8), 117.27 (C-6), 98.50 (C-3a) and 97.66 (C-9b); MS: m/z (EI) 217 (M⁺); Anal. Calcd. for C₁₁H₇NO₂S : C,60.81; H,3.24; N,6.44; S,14.75. Found: C,60.89; H,3.45; N,6.39; S,14.69.

3-Amino-4,5-dihydrothieno[3,4-c]quinolin-4-one (3c): Compound 3c was obtained as gray crystals (83%); m.p. 258°C; IR: 3405 and 3290 (NH₂) and 1635 cm⁻¹ (ring CO). H NMR: δ 6.79 (s, 1H, 1-H), 7.02-7.76 (m, 4H, Ar-H), 7.85 (br, 2H, NH₂) and 10.40 ppm (br, 1H, NH); ¹³C NMR: δ 163.73 (C-4), 160.88 (C-5a), 138.44 (C-3), 134.68 (C-9a), 129.56 (C-7), 128.48 (C-9), 125.58 (C-8), 121.88 (C-6), 117.84 (C-1), 105.79 (C-3a) and 103.96 (C-9b); MS: m/z (EI) 216 (M⁺); Anal. Calcd. for C₁₁H₈N₂OS (216.26): C,61.10; H,3.72; N,12.95. Found: C,61.60; H,3.81; N,12.93.

3-Aminothieno[3,4:3',4']naphtho[b]pyran-2-one (11): Compound 11 was obtained as brown crystals (82%) m.p. 302°C; IR:3400and 3280 (NH₂) and 1668 cm⁻¹ (ring CO). HNMR: δ 6.96 (s, 1H, 1-H) and 7.49-8.33 (m, 6H, Ar-H) and 10.42 ppm (br, 2H, NH₂); ¹³C NMR: δ 165.95 (C-2), 157.96 (C-11b), 145.03 (C-3), 132.62 (C-5), 130.77, 127.22, 126.28, 128.23, 122.63, 120.64, 120.31 and 112.45 (ring carbons), 97.44 (C-5a) and 96.96 (C-2a); MS: m/z (EI) 267 (M⁺); Anal. Calcd. for C₁₅H₉NO₂S (267.30): C,67.40; H,3.39; N,5.24; S,11.99. Found: C,66.96; H,3.76; N,5.15; S,12.27.

Reaction of (3b,c) and (11) with electron-poor olefins and acetylenes:

a) At 250°C: A mixture of each of 3b and 11 (0.01 mol) and diethyl fumarate or dimethyl acetylenedicarboxylate (0.01 mol) was heated at 250°C for 20 min. The reaction mixture was left to cool, then triturated with dioxane and precipitated by adding ethanol. The solid product, so formed, was collected by filtration and crystallized from the proper solvent.

Diethyl 4-amino-5-oxodibenzo[b,d]pyran-2,3-dicarboxylate (12a): Compound 12a was obtained as yellow crystals from dioxane/ethanol mixture (80%); m.p. 149°C; IR: 3405 and 3295 (NH₂), 1718 (ester CO) and 1695 cm⁻¹ (ring CO). HNMR: δ 1.19-1.43 (m, 6H, 2CH₃), 4.20-4.38 (m, 4H, 2CH₂), 7.38-7.56 (m, 4H, Ar-H) and 8.20-8.26 ppm (m, 3H, NH₂ and 1-H); ¹³C NMR: δ167.77 and 166.34 (ester CO), 161.56 (C-5), 152.37 (C-4), 151.53 (C-6a), 141.75 (C-2), 139.51 (C-10b), 138.9 (C-10a), 132.51 (C-8), 125.32 (C-9), 125.14 (C-10), 117.30 (C-7), 110.25 (C-4a), 107.62 (C-1), 105.03 (C-3), 62.04 and 61.81 (2 OCH₂), 14.27 and 14.05 (2 CH₃); MS: m/z (EI) 356 (M+1); Anal. Calcd. for C₁₉H₁₇NO₆ (355.34): C,64.22; H,4.82; N,3.94. Found: C,63.98; H,5.16; N,3.87.

Dimethyl 4-amino-5-oxodibenzo[b,d]pyran-2,3-dicarboxylate (12b): Compound 12b was obtained as light green crystals from dioxane/ethanol mixture (74%); m.p. 250°C; IR: 3505 and 3305 (NH₂), 1729 (ester

CO) and 1699 cm⁻¹ (ring CO). ¹ H NMR: δ 3.382 (s, 3H, CH₃), 3.87 (s, 3H, CH₃), 7.38 - 8.35 ppm (m, 7H, Ar-H and NH₂); ¹³C NMR: δ 168.00 and 167.71 (ester CO), 161.06 (C-5), 151.76 (C-4), 151.34 (C-6a), 144.22 (C-2). 140.87 (C-10b), 139.47 (C-10a), 132.07 (C-8), 124.87 (C-9), 124.64 (C-10), 116.92 (C-7), 116.83 (C-4a), 110.37 (C-1), 107.21 (C-3), 52.68 and 52.39 (2CH₃); MS: m/z (EI) 328 (M+1); Anal. Calcd. for C₁₇H₁₃NO₆ (327.29) : C,62.38; H,4.00; N,4.27. Found: C,61.91; H,4.15; N,4.24.

Diethyl 6-amino-2-oxonaphtho[1,2-b]benzo[d]pyran-4,5-dicarboxylate 13a: Compound 13a was obtained as light brown crystals from dioxane/ethanol mixture (75%); m.p. 178°C; IR: 3420 and 3310 (NH₂), 1710 (ester CO) and 1697 cm⁻¹ (ring CO). H NMR: δ 1.32 (t, 6H, 2CH₃, J = 7 Hz), 4.28 (q, 4H, 2OCH₂, J = 7 Hz) and 7.61-8.34 ppm (m, 9H, Ar-H and NH₂); 13 C NMR: δ 167.88, 166.35 (ester CO), 161.45(ring CO), 152.55 (C-6), 147.77 (C-1a), 142.00 (C-2a), 140.36(C-3), 134.81 (C-4), 124.96(C-5), 129.04, 128.30, 127.89, 122.84, 121.90, 120.99, 112.84, 108.03 and 104.91 (ring carbons), 62.07 and 61.81 (2 OCH₂), 14.30 and 14.09 (2 CH₃); MS: m/z (EI) 406 (M⁺); Anal. Calcd. for C₂₃H₁₉NO₆ (405.40): C,68.14; H,4.72; N,3.45. Found: C,67.72; H,4.55; N,3.41.

Dimethyl 6-amino-2-oxonaphtho[1,2-b]benzo[d]pyran-4,5-dicarboxylate 13b: Compound 13b was obtained as yellow crystals from dioxane/ethanol mixture (78%); m.p. 279°C; IR:3420 and 3310 (NH₂), 1730 (ester CO) and 1698 cm⁻¹ (ring CO). ¹ H NMR: δ 3.84 (s, 3H, OCH₃), 3.89 (s, 3H, OCH₃) and 7.61-8.31 ppm (m, 9H, Ar-H and NH₂); ¹³C NMR: δ 167.96, 166.48 (ester CO), 155.31 (ring CO), 141.40 (C-6), 140.17 (C-1a), 134.68 (C-2), 128.03(C-4), 124.75(C-5), 128.03, 127.62, 124.75, 121.75, 120.69, 107.92, 104.91 (ring carbons), 52.85 and 52.54 (2 OCH₃); MS: m/z (El) 378 (M+1); Anal. Calcd. for C₂₁H₁₅NO₆ (377.35): C,66.84; H,4.00; N,3.71.Found: C,66.40; H,4.09; N,3.65.

b) In Refluxing Dioxane: A mixture of each of 3b,c (0.01mol) and ethyl propiolate or dimethyl acetylenedicarboxylate (0.01mol) in dioxane was treated under reflux for four to twenty hours. The reaction mixture was evaporated under vacuo till half of its volume, the solid product, so formed, was collected by filtration and crystallized from the proper solvent.

Ethyl 4-amino-6-oxo-6H-thiepino [3,4-c]benzo [b]pyran-3-carboxylate (24a): Compound 24a was obtained as light green crystals in refluxing dioxane for 20 h., crystallized from DMF (80%); m.p. 286 °C; IR: 3415 and 3330 (NH₂),1720 (ester CO) and 1662 cm⁻¹ (ring CO). ¹ H NMR: δ 1.26 (t, 3H, CH₃, J =7 Hz), 4.19 (q, 2H, OCH₂, J =7 Hz), 5.90 (d, 1H, H-2, J = 14 Hz), 7.21-7.95 (m, 4H, Ar-H), 8.23 (d, 1H, H-1, J = 14 Hz) and 8.34 ppm (br, 2H, NH₂); ¹³C NMR: δ 166.23 (ester CO), 166.05 (C-6), 159.04 (C-4), 133.40, 135.33, 132.95, 130.43, 125.50, 124.80, 117.98, 117.70, 114.11 and 111.86 (ring carbons), 110.50 (C-3), 59.85 (OCH₂) and 14.22 (CH₃); MS: m/z (EI) 315 (M⁺); Anal. Calcd. for C₁₆H₁₃NO₄ S (315.27): C,60.95; H,4.16; N,4.44.Found: C,60.73; H,4.20; N,4.48.

Dimethyl 4-amino-6-oxo-6H-thiepino[3,4-c]benzo[b]pyran-2,3-dicarboxylate (24b): Compound 24b was obtained as red crystals in refluxing dioxane for 4 h., crystallized from ethanol (80%); m.p. 162° C; IR: 3425 and 3330 (NH₂),1720 (ester CO) and 1695 cm^{-1} (ring CO). HNMR: δ 3.58 (s, 3H, OCH₃), 3.70 (s, 3H, OCH₃), 7.28 (s, 1H, H-1), 7.30-7.46 (m,4H, Ar-H) and 7.96 ppm (br, 2H, NH₂); MS: m/z (EI) 359 (M[†]); Anal. Calcd. for C₁₇H₁₃NO₆ S (359.27): C,56.83; H,3.65; N,3.90.Found: C,56.66; H,3.63; N,3.50.

Ethyl 4-amino-6-oxo-6,7-dihydro-thiepino[3,4-c]quinoline-3-carboxylate (24c): Compound 24c was obtained as brown crystals in refluxing dioxane for 20 h., crystallized from DMF (77%); m.p. 323°C; IR: 3360 and 3290 (NH₂),1690 (ester CO) and 1651 cm⁻¹ (ring CO). ¹H NMR: δ 1.26 (t, 3H, CH₃, J = 7 Hz), 4.18 (q, 2H, OCH₂, J = 7 Hz), 5.75 (d, 1H, H-2, J = 14 Hz), 7.17-7.94 (m, 4H, Ar-H), 8.22 (d, 1H, H-1, J = 14 Hz), 8.34 (br, 2H, NH₂) and 10.73 ppm (br, 1H, NH); ¹³C NMR: δ 166.49 (ester CO), 163.65 (C-6), 160.98 (C-4), 138.90, 136.54, 136.20, 129.56, 128.03, 125.58, 123.33, 121.67, 117.59, 110.22 and 107.10 (ring carbons), 59.59 (OCH₂) and 14.30 (CH₃); MS: m/z (EI) 314 (M⁺); Anal. Calcd. for C₁₆H₁₄N₂O₃ S (314.29): C,61.14; H,4.49; N,8.91.Found: C,60.93; H,4.63; N,9.12.

Dimethyl 4-amino-6-oxo-thiepino[3,4-c]dihydroquinoline-2,3-dicarboxylate (24d): Compound 24d was obtained as red crystals in refluxing dioxane for 4 h., crystallized from ethanol (74%); m.p. 248°C; IR: 3425 and 3330 (NH₂),1720 (ester CO) and 1695 cm⁻¹ (ring CO). ¹ H NMR: δ 3.58 (s, 3H, OCH₃), 3.65 (s, 3H, OCH₃), 7.04 (s, 1H, H-1), 7.20-7.38 (m,4H, Ar-H), 7.69 (br, 2H, NH₂) and 10.61 (br, 1H, NH); ¹³C NMR: δ 166.27 and 164.96 (ester CO), 162.66 (C-6), 161.60 (C-4), 138.17, 137.52, 135.40, 131.17, 128.90, 125.58, 124.75, 121.98 and 117.99 (ring carbons), 116.48 (C-3), 115.62 (C-2), 53.54 and 52.26 (2OCH₃); MS: m/z (EI) 358 (M⁺); Anal. Calcd. for C₁₇H₁₄N₂O₅ S (358.27): C,56.98; H,3.94; N,7.82, S,8.93. Found: C,56.60; H,4.06; N,7.56; S,9.15.

Dimethyl 4-amino-2-oxo-thiepino[3,4-c]naphtho[1,2-b]pyran-5,6-dicarboxylate (25) :Compound 25 was obtained as red crystals in refluxing dioxane for 4 h., crystallized from ethanol (84%); m.p.233°C; **R**: 3420 and 3315 (NH₂),1730 (ester CO) and 1712 cm⁻¹ (ring CO). ¹ H NMR: δ 3.55 (s, 3H, OCH₃), 3.72 (s, 3H, OCH₃), 7.56 (s, 1H, H-7), 7.45-7.87 (m, 6H, Ar-H) and 8.32 ppm (br, 2H, NH₂); ¹³C NMR: δ166.67 and 165.23 (ester CO), 164.71 (C-2), 161.58 (C-4), 158.68, 148.20, 142.31, 136.38, 133.46, 133.06, 132.94, 127.96, 127.82, 127.62, 124.16, 123.57, 122.00, 121.25 and 113.53 (ring carbons), 53.71 and 52.41 (2 OCH₃); **MS**: m/z (EI) 409 (M⁺); Anal. Calcd. for C₂₁H₁₅NO₆S (409): C,61.61; H,3.69; N,3.42, S,7.81. Found: C,61.83; H,3.76; N,3.21; S,7.40.

Reaction of (3b,c) and (11) with N-phenylmaleimide: A mixture of each of 3b,c and 11 was fused with N-phenylmaleimide at 250°C for 5 min., then left to cool at room temperature and triturated with ethanol. The solid product, so formed, was collected by filtration and crystallized from dimethyl formamide.

Compound 26a was obtained as yellow powder (86%); m.p.>350°C; IR: 3430 and 3353 (NH₂), 1752 (ring CO) and 1699 cm⁻¹ (amide CO); MS: m/z (EI) 357 (M+1); Anal. Calcd. for $C_{21}H_{12}N_2O_4$ (356.32): C,70.78; H,3.39; N,7.86. Found: C,70.20; H,3.39; N,7.76.

Compound 26b was obtained as orange powder (85%); m.p.>350°C; IR: 3430 and 3280 (NH₂), 1739 (amide CO) and 1694 cm⁻¹ (ring CO); H NMR: δ 4.86 (br, 2H, NH₂), 7.27-7.68 (m, 8H, Ar-H), 7.99 (s, 1H, H-1), 8.45 (d, 1H, H-1, J=7.6 Hz) and 10.45 ppm (br, 1H, NH); MS: m/z (EI) 356 (M+1); Anal. Calcd. for $C_{21}H_{13}N_3O_3$ (355.09): C,70.98; H,3.69; N,11.82. Found: C,70.98; H,3.76; N,11.83.

Compound 27 was obtained as green powder (76%); m.p.>350°C; IR: 3445 and 3335 (NH₂), 1754 (ring CO) and 1694 cm⁻¹ (amide CO); ¹ H NMR: δ 7.30-8.60 ppm (m, 14H, Ar-H); MS: m/z (EI) 407 (M+1); Anal. Calcd. for $C_{25}H_{14}N_{2}O_{4}$ (406.40): C.73.89; H.3.47; N.6.89. Found: C.73.99; H.3.46; N.6.86.

Reactions of 1a,b:

- a) With ethyl acrylate: A mixture of 1b (3.45g, 0.01mol) and ethyl acrylate (1.0g; 0.01mol) was refluxed in dioxane acetic acid 1:1 (10 mL) for 24h then the solvent was evaporated under vacuo, the solid product, so formed, was collected by filtration and crystallized from acetic acid. Compound 31b was obtained as red crystals (76%); m.p.206°C; **IR:** 3500 and 3400 (NH₂), 1715 (thiolaceton) and 1675 (ester CO) and 1620 cm⁻¹ (ring CO); ¹ **H NMR:** δ 1.22 (t, 3H, CH₃, J=7.0 Hz), 1.80 (s, 3H, CH₃), 3.32 (s, 2H, CH₂), 3.80 (s, 3H, OCH₃), 4.20 (q, 2H, CH₂, J=7.0 Hz), 7.0-7.50 (AA'BB', 4H, Ar-H) and 8.0 ppm (br, 2H, NH₂); Anal. Calcd. for C₂₀H₁₀N₃O₅S (413.44): C,58.10; H,4.63; N,10.16; S,7.75. Found: C,57.94; H,4.90; N,10.08; S,8.03.
- b) With diethyl fumarate: A mixture of 1a (3.15g, 0.01mol) and diethyl fumarate (1.72g, 0.01mol) was heated at 250°C for 5 min., left to cool, then triturated with ethanol and crystallized from ethanol. Compound 31c was obtained as red crystals (75%); m.p.206°C; IR: 3505 and 3295 (NH₂), 1714 (ester CO) and 1673 cm⁻¹ (ring CO). ¹ H NMR: δ 1.01-1.30 (m, 6H, 2CH₃), 1.96 (s, 3H, CH₃), 3.95-4.00 (m, 4H, 2 OCH₂), 4.24 (s,1H, H-5), 7.33-7.58 (m, 5H, Ar-H), 8.40 ppm (br, 2H, NH₂); MS: m/z (EI) 455 (M⁺); Anal. Calcd. for $C_{22}H_{21}N_3O_6S$ (455.11): C,58.01; H,4.65; N,9.23; S,7.04. Found: C,57.83; H,4.65; N,9.17; S,7.08.
- c) With N-phenylmaleimide: A mixture of 1a (3.15g, 0.01mol) and N-phenylmaleimide (1.73g, 0.01mol) was heated at 250°C for 5 min., left to cool, then triturated with ethanol and crystallized from dioxane. Compound 32 was obtained as yellow crystals (72%); m.p.287°C; IR: 3450 and 3350 (NH₂), 1738 (ester CO), 1695 (amide CO) and 1642 cm⁻¹ (ring CO). ¹ H NMR: δ 2.55 (s, 3H, CH₃), 3.95 (s, 3H, OCH₃), 7.47-7.53 (m, 10H, Ar-H) and 8.69 ppm (br, 2H, NH₂); ¹³C NMR: δ 166.34 (C-6) and (C-8), 164.90 (ester CO), 159.89 (C-4), 146.51 (C-1), 140.45 (C-5), 131.75 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.4, 128.65, 127.82 and 127.02 (ring carbons), 125.87 (C-9), 135.40 (C-9), 135.4

9a), 125.77 (C-4a), 119.07 (C-8a), 117.31 (C-5a), 53.22 (C-OCH₃), 12.09 and 11.15 (CH₃); MS: m/z (EI) 455 (M⁺); Anal. Calcd. for $C_{25}H_{18}N_4O_5$ (454.43): C,66.07; H,3.99; N,12.33. Found: C,66.24; H,3.99; N,12.69

REFERENCES:

- Gaertner R. and Tonkyn R.G. J. Am. Chem. Soc. 1951, 73, 5872; Clapp D.B. J. Am. Chem. Soc. 1939, 61, 2733.
- Kotsuki H.H.; Kitagawa S.; Nishizawa H. and Tokoroyama T. J. Org. Chem. 1978, 43, 1471. Kotsuki H.H.; Nishizawa H.; Kitagawa S.; Ochi M.; Yamasaki N.; Matsuoka K. and Tokorayama T. Bull. Chem. Soc. Jpn. 1979, 52, 544.
- 3. Jursic B.S. J. Heterocyclic Chem. 1995, 32, 1499; ibid 1995, 32, 1445.
- 4. Elnagdi M.H.; Negm A.M. and Sadek K.U. Synlet, 1994, 27.
- 5. Elnagdi M.H. and Erian A.W. Liebigs, Ann. Chem. 1990, 12515.
- 6. Elnagdi M.H.; Negm A.M. and Erian A.W. Liebigs, Ann. Chem. 1989, 1255.
- 7. Elnagdi M.H.; Negm A.M.; Hassan E.M. and El-Borei A. J. Chem. Res (S) 1993, 130.
- 8. Al-Awadhi H.; Al-Omran F.; Elnagdi M.H.; Infantes L.; Foces-Foces C.; Jagerovic N. and Elguero J., Tetrahedron 1995, 51, 12745.
- 9. Gronowitz, S. and Dahlgren, T. Chem. Scripta 1977, 12, 57.
- 10. Nyiondi-Bonguen E.; Sophue Fondjo E.; Tanee Fomum Z. and Döpp D. J. Chem. Soc. Perkin Trans. 1, 1994, 2191.
- 11. Dewar, M.J.S. and Trinajstic, N. J. Amer. Chem. Soc. 1970, 92, 1453.
- 12. Barton, T.J.; Martz, M.B. and Zika, R.G. J. Org. Chem. 1972, 1232.
- 13. Schlessinger, R.H. and Ponticello, G.S. Tetrahedron Lett. 1968, 3017, ibid, 1969, 4361.
- 14. Reinhouldt, D.N. and Kouwenhoven, C.G. J. Chem. Soc. Chem. Commun. 1972, 1232.
- 15. Hofmann, H.; Meyer, B. and Hofmann, P. Angew. Chem. 1972, 84, 477.
- 16. Hofmann, H. and Westernacher, H. Chem. Ber. 1969, 102, 205.

(Received in UK 26 April 1996; revised 23 July 1996; accepted 25 July 1996)